Source Test Procedure **ST-2**

BERYLLIUM

(Adopted January 20, 1982)

REF: Regulation 11-3-301

1. APPLICABILITY

1.1 This procedure is used to quantify emissions of Beryllium. It determines compliance with Regulation 11-3-301.

2. PRINCIPLE

2.1 Particulate matter (including Beryllium) is withdrawn isokinetically from the stack gas stream and collected on filters in the stack. The weight of the Beryllium collected is measured by atomic absorption spectrophotometry according to Analytical Procedure Lab-2.

3. RANGE

3.1 The minimum measurable emission of Beryllium is .001 gr/SDCF.

4. INTERFERENCES

None Known

5. APPARATUS

- 5.1 Probe Nozzle. The sampling train and its components are shown in Figure 2-1. The probe nozzle shall be constructed of borosilicate glass, quartz, or stainless steel.
- 5.2 Filter Medium. Use Millipore Type "A" glass fiber disc type or equivalent.
- 5.3 Connections. The connection between the filters and the first impinger must be able to withstand stack temperatures. Vinyl tubing is acceptable in making all other connections.
- 5.4 Pitot Tube. Use a Stauscheibe (Type-S), or equivalent, with a known coefficient which is constant within \pm 5% over the entire working range. The pitot tube coefficient is determined by placing both the S-type and the standard pitot tube in a gas stream and measuring the pressure head with both over the entire velocity range of interest. Calculate the coefficient of the Type-S pitot tube as follows:

$$Cp_{s} = Cp_{std} \left[\frac{\Delta P_{std}}{\Delta P_{s}} \right]^{\frac{1}{2}}$$

where:

Cp_S = Type-S pitot tube coefficient

 ΔP_{S} = Pressure head, Type-S pitot tube

- ΔP_{std} = Pressure head, standard pitot tube
- Cp_{std} = Standard pitot tube coefficient
- 5.5 Temperature Measuring Device. Use a Chromel-Alumel thermocouple accurate to \pm 15°F, connected to a temperature compensated null type potentiometer, or equivalent, to measure stack temperatures.
- 5.6 Condensers. Use three Greenberg-Smith impingers. The third impinger shall be modified by removing the impaction plate and attaching a thermometer to the inlet stem.
- 5.7 Cooling System. Use an ice bath to contain the impingers.

Figure 2-1



Beryllium Sampling Train

- 5.8 Sample Pump. Use a leak-free vacuum pump capable of maintaining a 1.0 CFM flow rate at 15 inches of mercury. The pump must have a sample rate control valve and a vacuum gauge attached to the inlet.
- 5.9 Silica Gel Tube. Use approximately 500 cc of silica gel, followed by a Drierite indicator, to insure that the gas entering the dry test meter is free of H_2O .
- 5.10 Dry Test Meter. Use a 175 CFH dry test meter accurate within \pm 2% of the true volume and equipped with a thermometer to measure the outlet temperature. The working pressure across the meter shall not exceed a one inch water column.
- 5.11 Rotameter. Use a calibrated rotameter to measure the sampling rate.

- 5.12 Pressure Gauge. Use a Magnehelic differential pressure gauge, or equivalent, in the same range as the velocity and static pressures being measured in the stack.
- 5.13 Analytical Balance. An analytical balance capable of measuring condensate weights to the nearest 0.1 gram is acceptable.
- 5.14 Barometer. Use a barometer that is accurate to within \pm 0.2 inches of mercury.

6. PRE-TEST PROCEDURES

- 6.1 Impinger Preparation. Fill each of two unmodified Greenberg-Smith impingers with approximately 100ml of distilled water. Weigh and record the weights on the data sheet as shown in Form 2-1.
- 6.2 Nozzle Size Determination. (Same as Section 6.3.4 in ST-15)
- 6.3 Assemble the sampling train as shown in Figure 2-1.
- 6.4 The entire sampling train must be leak-checked before each test run. Plug the sampling probe, start the pump, and adjust the pump vacuum to 380 mm Hg (15" Hg). A leak rate through the meter which exceeds 0.57 lpm (0.02 CFM) is unacceptable.

7. SAMPLING

- 7.1 Each test run shall be of 50 minute duration when testing emissions from continuous operations. Each test run at batch process operations shall be for 90% of the batch time or 50 minutes, whichever is less.
- 7.2 When inserting the probe into the stack rotate the nozzle so it points down stream to avoid a particulate collection prior to sampling. Immediately before sampling rotate the probe so that nozzle points upstream.
- 7.3 Sample at the traverse points determined in acc ordance with ST-18.
- 7.4 Record the following information at five-minute intervals or whenever changing sampling locations on a field data sheet as shown in Form 2-2.

Stack velocity head Sample time Isokinetic sample rate Cumulative sample volume Impinger saturation temperature Stack gas temperature Impinger vacuum Dry test meter temperature

- 7.6 Add ice as necessary to maintain impinger temperatures at 7 °C (45°F) or less.
- 7.7 At the conclusion of each run, stop the pump, remove the probe from the stack and record the final meter reading. Point the probe upward and purge the sample train with ambient air.
- 7.8 Conduct three consecutive test runs.

8.

8.1 Analyze the filters and any material in the nozzle for Beryllium according to Analytical Procedure Lab 2.

9. AUXILIARY TESTS

9.1 Determine the CO₂, O₂ and CO concentrations simultaneously with each particulate run in accordance with ST-5, ST-14 and ST-6.

10. CALCULATIONS

10.1 Standard Dry Sample Volume:

$$V_{o} = \frac{17.71V_{m}P_{b}}{T_{m}}$$

Where:

- V_0 = Standard dry sample volume, SDCF at 70 0 F and 29.92 inches Hg
- V_m = Actual metered volume, ft³
- P_b = Barometric pressure, inches Hg

 T_m = Average meter temperature ^OR

17.71 = Constant correcting to 70° F and 29.92 inches H₂O

10.2 Water Vapor Content

$$H_{2}O = \frac{(0.0474 \, W_{c}) + \frac{V_{o}P_{sat}}{P_{b} - P_{i} - P_{sat}}}{V_{o} + (0.0474 \, W_{c}) + \frac{V_{o}P_{sat}}{P_{b} - P_{i} - P_{sat}}} \times 100$$

Where:

P_{sat} = Water saturation pressure, inches Hg

P_b = Barometric pressure, inches Hg

P_i = Pump inlet vacuum, inches Hg

- H₂O = Percent water vapor
- 0.0474 = Cubic feet of vapor resulting from 1 cubic centimeter of liquid H_2O

10.3 Stack Gas Molecular Weight

$$MW = 0.44 (\%CO_2) + 0.32 (\%O_2) + 0.28 (\%N_2 + \%CO) + 0.18 (\%H_2O)$$

Where:

MW	=	Molecular Weight
%CO ₂	=	Percent Carbon Dioxide by volume(dry basis)
%0 ₂	=	Percent Oxygen by volume (dry basis)
%CO	=	Percent Carbon Monoxide by volume (dry basis)
%H ₂ O	=	Percent Moisture by volume

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- %N₂ = Percent Nitrogen by volume (dry basis determine by difference)
- 10.4 Stack Gas Flow Rate Determine in accordance with ST-17.
- 10.5 Isokinetic Ratio. Calculate for each traverse point as:

$$R_{i} = \frac{T_{si} Q_{mi}}{60(100 - H_{2}O)AV_{si}t_{i}T_{mi}} \times 100\%$$

Where:

Ri	=	Isokinetic ratio at given point
ti	=	Time, at point i, minutes
А	=	Nozzle area, ft ²
V _{si}	=	Stack velocity, point i, FPS
T _{mi}	=	Meter temperature, point i, ${}^{\rm O}{\rm R}$
T _{si}	=	Stack temperature, point i, ${}^{\rm O}{\rm R}$
Qm	i=	Metered volume, point i
60	=	Minutes/hr.

Overall isokinetic ratio for each run:

$$R = \frac{Q_{mi}}{60(100 - H_2 O) A T_m V_{si} t_i T_{si}} \times 100\%$$

10.6 Mass emissions. the emission rate of Beryllium shall be calculated as:

$$M = \frac{W \times Q_{o} \times 60 \times T}{Vo}$$

Where:

M = Mass emission rate, g/day

W = Total weight of Be in filter and nozzle, grams (g)

Q₀ = Stack gas flowrate, SDCFM

V = Sample Volume, SDCF

T^O = Plant operation, hr/day

60 = minutes/hour

11. **REPORTING**

11.1 The data and information indicated in Form 2-3 shall be reported.

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Form 2-1

Source Test Laboratory Data Sheet

Impinger Weightings

Plant Name:	Plant Number:
Source Operation:	Test Date
Source Test #:	Page: 1 of
Impinger Solution:	Initial:

Impinger	(A)	(B)	(C)
I. D. #	Tare Weight (g)	Filled Weight (g)	Final Weight (g)

Impinger I. D. #	(C-A) Sample Weight (g)	(C-B) Condensate Wt. (q)	Condensate Weight / Run (g)
			Run A
			Run B
			Run C
			Run D

Plant #	Bay Are	a Air	Quality	y Mar	nageme	nt Disti	rict		Nozzle [Diameter
Source I.D.	939 Ellis Street, San Francisco, CA 94109						Pitot Tube I.D., Cp			
Sample Type	L		For	m 2-2	<u>)</u>				Gas Sys	stem
Process Cycle		Sou	rce Te	st Dat	a Sheet				– Pbar, Ba	arometer
Duct Size									Leak Te	st Rate
Duct Shape	Run #			_ [Date:				Time @	Point
Duct Pressure	Temp Meter #			Box	ΔH@		<u> </u>		# of Poir	nts
Assumed %H ₂ 0	Mag. Gauge #			_ Met	er (Y)				Time/Ru	ın (Min.)
Sampling Train: Probe #	Filter #		Imp. #		Imp	o. #	P	ump/Box # _		
Initial Traverse Data					Sam	oling Data	l			
$\begin{array}{c cccc} Trav. & Dist. & Duct & \Delta P & Ar \\ Point & from & Temp. & "H_20 & o \\ I.D. & Wall & {}^{\circ}F & & Fl \end{array}$	ngle Traverse of Point low I.D.	ΔΡ "H ₂ 0	Duct Temp. °F	Vs FPS	Time (minutes)	Meter Rate CFH	Meter Temp. °F	Meter Volume Ft ³	Train Vacuum "Hg	Sat'd Gas Temp. °F
Post Run Impinger Catch (ml) =		Source	Test Tear	n	Cor	nments:				
$Assumed CO_{2} = $										

			F	Form 2-3						
Distribut Firm Permit S Enforcer Technica Planning Request DAPCO	tion: Bervices ment Services al Services g ter	AI	BAY AREA R QUALITY MANAGEMENT DISTRICT 939 Ellis Street San Francisco, California 94109 (415) 771-6000 Summary of Source Test Results					Report No.:		
	S	ource Ir	formation				BAAQI	ID Repres	entat	ives
Firm Name and Address			Firm Representative and Title Source Test Engineers				ngineers			
Permit Con	ditions:		Source:			Perm	it Service	s Division/Enfo	rcemer	nt Division
			Plant No. Permit No. Test Reques Operates		Requeste	ted By:				
Operating F	Parameters:									
Applicable	e Regulations:					VN Recommended:				
Source Te <u>METHOD</u>	est Results and TEST	Comment	is:	<u>RUN A</u>	<u>RUI</u>	<u>N B</u>	<u>RUN (</u>	<u>AVERA</u>	<u>GE</u>	<u>LIMIT</u>
ST-17 ST-23 ST-14 ST-5 ST-6	Stack Volume Stack Gas Te Water Conter Oxygen, Volu Carbon Dioxie Carbon Mono Carbon Mono	e Flowrate emperatur nt, Volume ime % de, Volum oxide, ppn oxide, lb/h	e, SDCFM re, °F e % ne % nv r							
ST-2	ST-2 Beryllium Emissions, gm/day Isokinetic Ratio, act./theo.									

Air Quality Engineer II Date		Supervising Air Quality Engineer	Date	Approved by Air Quality Engineering Manager